

Review on SmFeO3 prepared by different synthesis techniques

Preksha Gangneja¹, Sarita Khaturia², Narender Kumar^{1**}, Shalendra Kumar^{3***} and Kavita Kumari^{1*}

¹Department of Physics (SLAS), Mody University of Science and Technology, Sikar Rd, Lakshmanagarh, Narodara Rural, Rajasthan 332311, India

²Department of Chemistry (SLAS), Mody University of Science and Technology, Sikar Rd, Lakshmanagarh, Narodara Rural, Rajasthan 332311, India

³Department of Physics, University of Petroleum & Energy Studies, Dehradun 248007, India

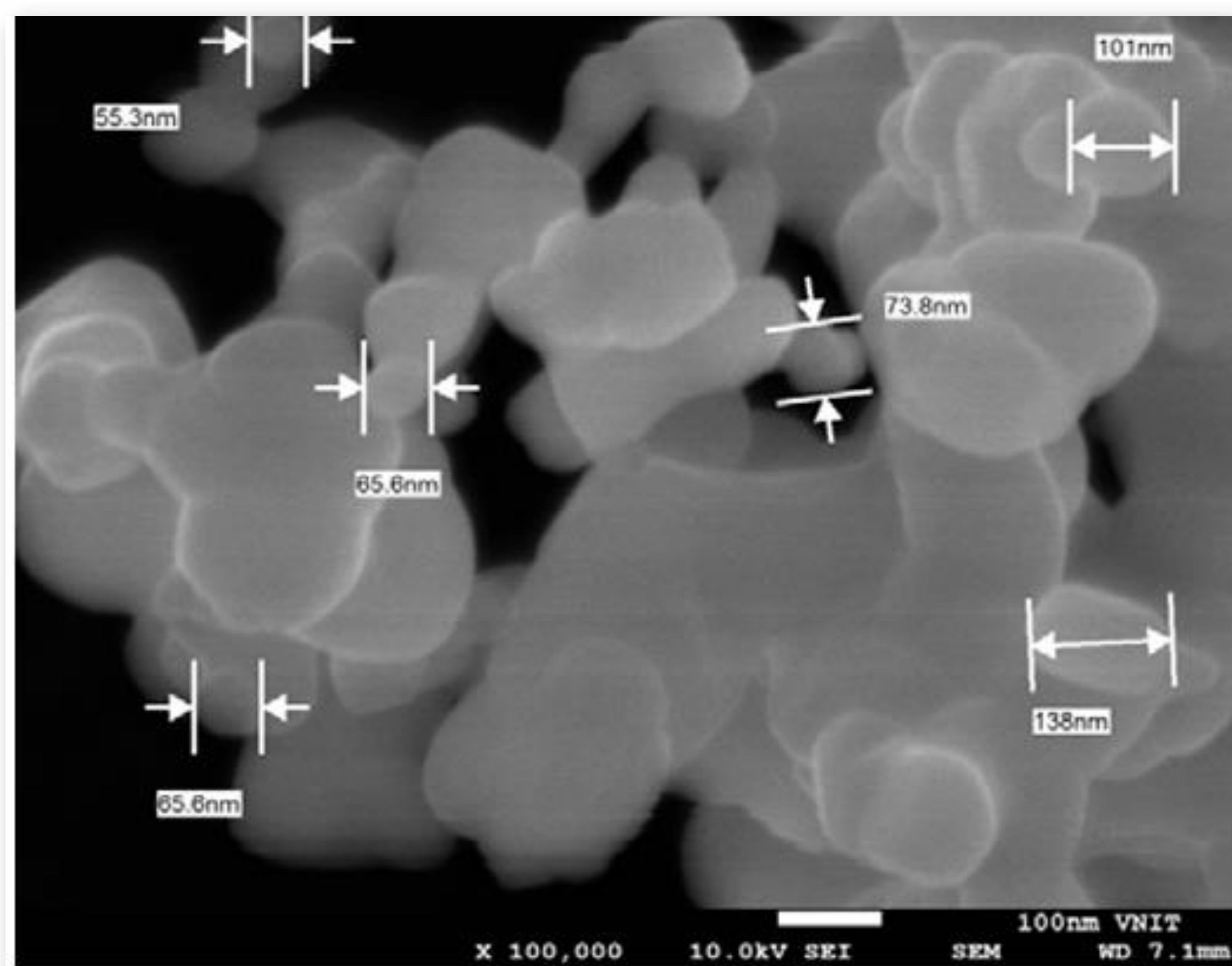
Corresponding Author – *kkmalhan@gmail.com, **narendrakumar.slas@modyuniversity.ac.in, ***shailuphy@gmail.com

Abstract

Samarium Orthoferrite (SmFeO₃) is an orthorhombic perovskite structured material with significant applications in the area of magnetic sensors & energy storage. The synthesis method plays a crucial role for determining the structural, morphological and magnetic properties. This review compares three widely used synthesis techniques; sol-gel method, solid-state reaction and ball-milling method. The relative effect of these methods on the structure of the compound and its various properties have been investigated through a vast literature review. For instance, the influence of the method on the sizes of the nanostructures has been found to be 46nm for the sample prepared by the sol-gel method, whereas the solid-state reaction method results in 63 nm. Similarly, the ball-milling method provides a top-down approach for size reduction & strain-induced defect. Thus, a comparative analysis of these materials reveals variation in optical bandgaps, structural integrity & functional properties of SmFeO₃.

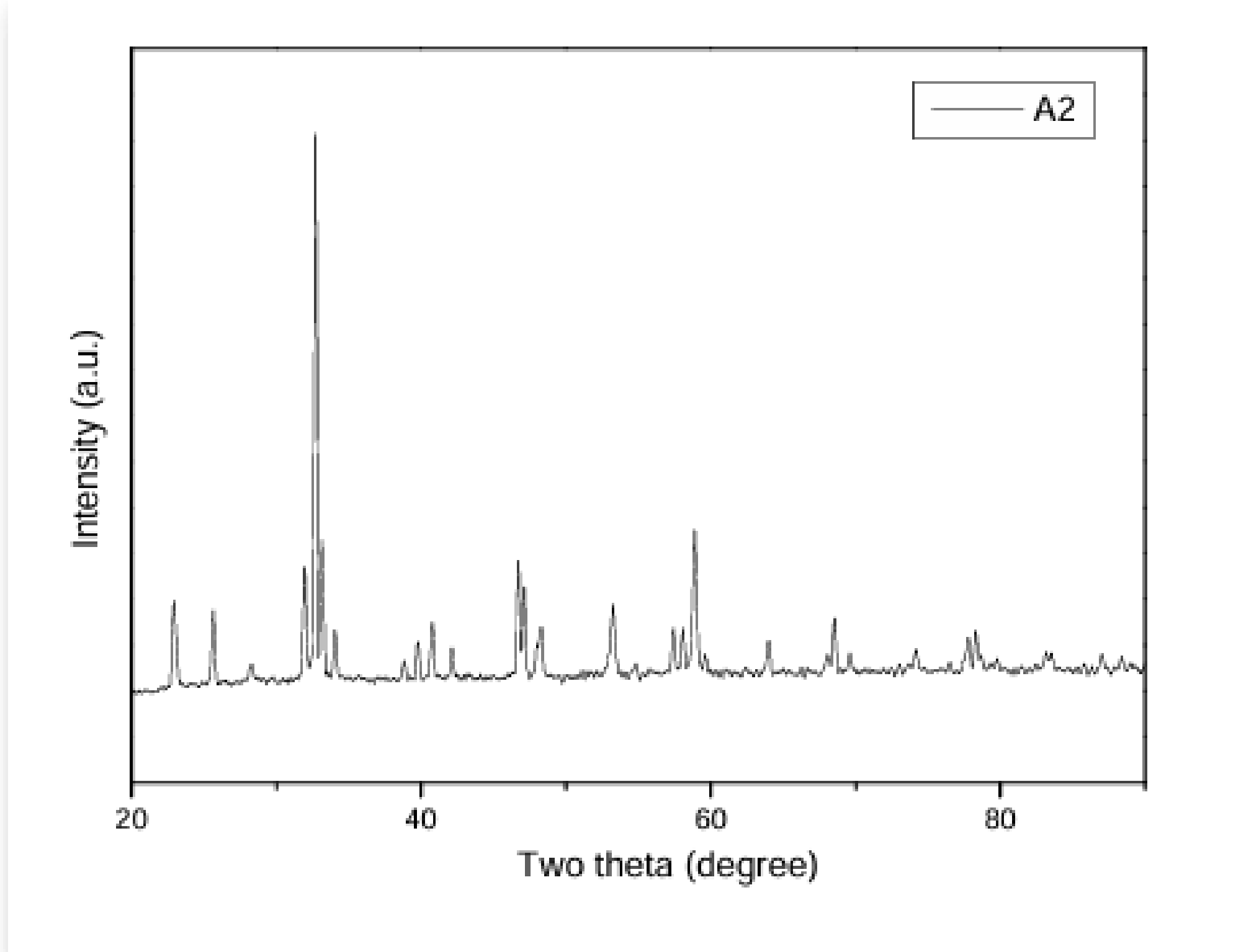
Keywords: SmFeO₃, Sol-gel method, Solid-state reaction and Ball-milling method

Literature Review



- SEM indicates the particle size are irregular until its average size is 100 nm
- Sol-gel method has 46nm crystalline size due to lower-reaction temperature meanwhile, solid-state reaction has 63 nm crystallite size due to prolonged diffusion & grain growth.

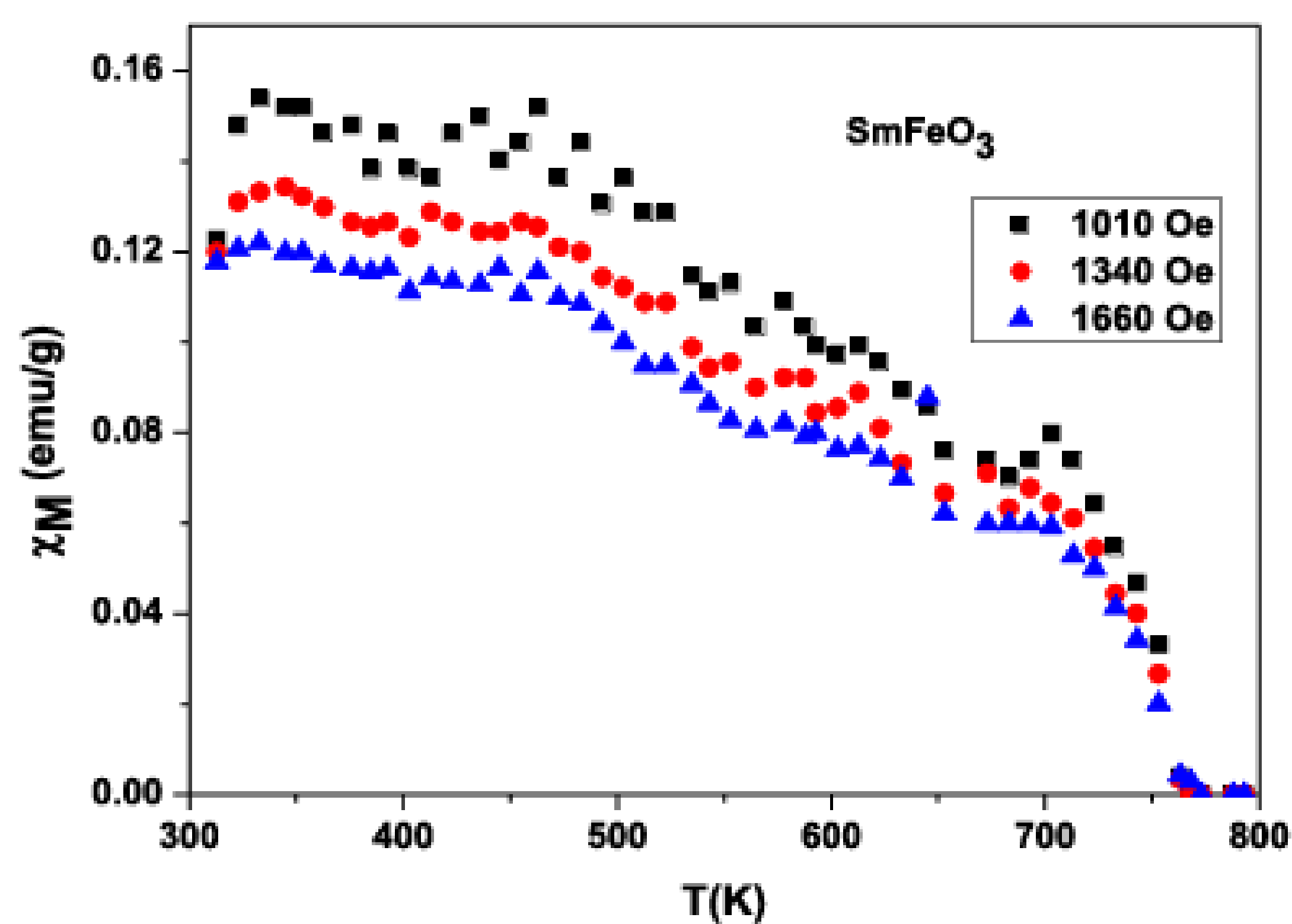
- XRD pattern indicates the single phase orthorhombic structure with confirms that they belong to space group Pnma



- The relative intensity of the XRD pattern is observed as milling time increases which might indicate the decrease in crystal size

- The relationship between χ_m and the absolute temperature at different magnetic field strength for SmFeO₃ nanoparticles

- χ_m values rise with increasing temperature until $T_{sr} = 463K$ when they begin to fall slowly until they reach the Neel temperature (T_n)

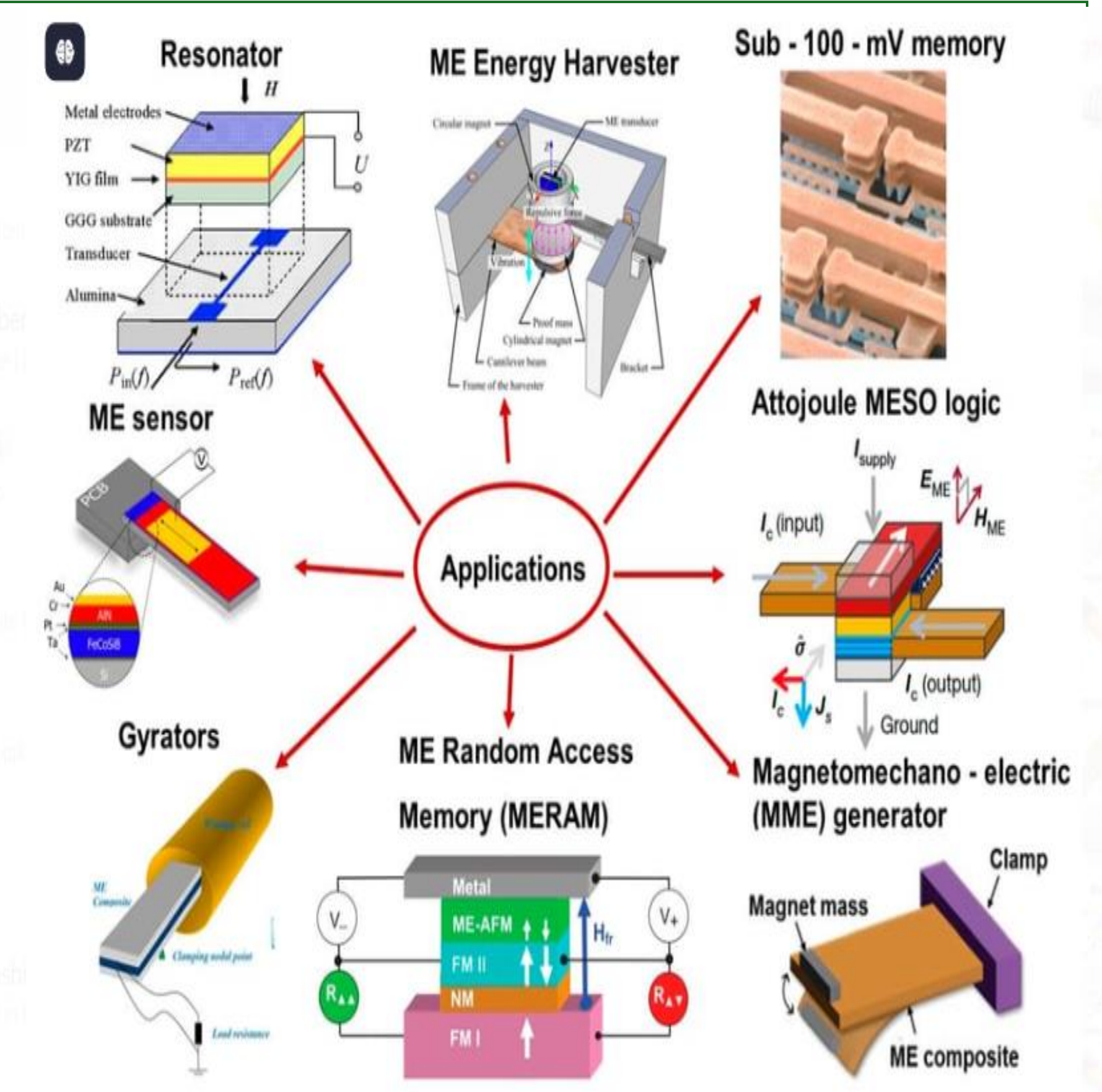
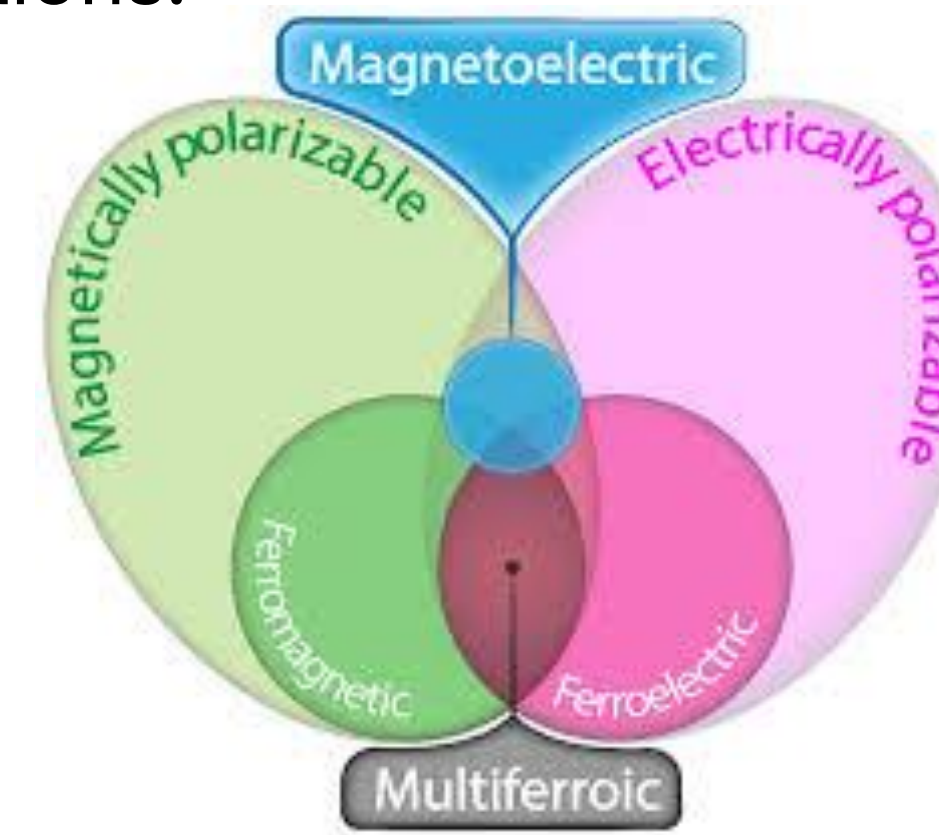


- After Neel temperature, sample loses its anti-ferromagnetic properties and begins to exhibit paramagnetic behaviour

Introduction

Samarium Orthoferrite (SmFeO₃) is a perovskite structure its general formula is ABX₃ or ABO₃. It is used in widely applications.

1. Resonator
2. ME Energy
3. Gytrators
4. ME Random Access memory
5. ME sensor
6. Sub-10 mv memory
7. Magnetomechano-electric (MME) generator
8. Attojoule MESO logic



And its many properties such as multiferroic, catalyst, gas sensors etc.

Different synthesis techniques

Samarium orthoferrite (SmFeO₃) has rare-earth orthoferrite that has gained significant attention due to its unique structural, magnetic & optical properties. It belongs to the perovskite family (RFeO₃ where, R = rare earth metal)

The perovskite SmFeO₃ are highly dependent on its synthesis techniques as different methods influence the particle size, crystallinity & morphology. Various synthesis techniques including sol-gel method, solid-state reaction and ball-milling method have been employed to optimize physical & chemical properties.

Each technique offers distinct advantages in terms of phase-purity, size reduction and magnetic properties.

This review provides in-depth comparison of different synthesis techniques highlighting the effect of structural, optical and magnetic properties.

Conclusion

- XRD pattern indicates that orthorhombic structure or Pnma (or space group)
- Ball-milling method, with increasing milling time, decreases crystallite size
- By sol-gel method, crystallite size is 46nm, while solid-state reaction results in a crystallite size of 63 nm because the sol-gel method has a lower reaction temperature, while solid-state reaction is due to prolonged diffusion & grain growth.
- Magnetic ordering of SmFeO₃ phases changes from anti-ferromagnetic to ferromagnetic due to thermal disorder.

References

- [1] R B Mankar, ' Synthesis of SmFeO₃ perovskite oxide by sol-gel method'.
- [2] M.M Arman, 'Novel multiferroic nanoparticles Sm_{1-x}Ho FeO₃ as a heavy metal Cr⁶⁺ removal from water.